

7 Eastgate Dr. • P.O. Box 790 • Jacksonville, IL 62651-0790 217-245-2183 • Fax: 217-243-7634 • www.ilmoproducts.com

# **Certificate of Analysis**

Certificate ID:

15567

Part #:

BAC105L080T

Cylinder Size:

105L 06723080A5

Lot Number: **Expiration:** 

4/5/2025

0.080 BAC (For the calibration of instruments used to determine breath alcohol concentration)

Contents:

105 Liters @ 1000 psig 70°F (21°C)

**Analytical** 

Reported

**Accuracy** 

**Analytical** 

Component:

**Concentration:** 

Method: (U, k=2):

+/-0.002 BAC(G/210L) NDIR

Ethanol Nitrogen 208 ppm Balance

[5.2 ppm]

316 East Ninth Street Owensboro, KY 42303 Phone 866-835-0690

Distributed by:

CMI Inc.

www.alcoholtest.com

\*Traceable to:

Certified Reference Material - 261.0 µmol/mol

Ethanol in Nitrogen - Serial No. ND7017 Lot No. 080722E2

Store in dry area, away from sources of heat, ignition and direct sunlight. Do not allow storage area to exceed 52 °C (125 °F).

Specialty Gas Lab Tech

Accreditation #61895

The calibration results within this certificate were obtained at the facility listed above using equipment and standards capable of producing analytical results traceable to NIST, and apply only to the items contained on this certificate. ILMO Products Company makes no warranty or representation as to the suitability of the use of any information provided for any particular purpose. The information use is at the sole discretion and risk of the user. Liability shall be limited to established replacement cost of this material or service. This certificate applies only to the items described and shall not be reproduced other than in full, without written approval from the issuing facility.



# **Safety Data Sheet**

Ethanol in Nitrogen

www.limoproducts.com

#### Section 1: Product and Company Identification

ILMO Products Company 7 Eastgate Drive, Jacksonville, Illinois 62650 217-245-2183 800-424-9300 (Chemtrec)

Fax 217-243-7634 E-mail: info@ilmooroducts.com Web: www.ilmoproducts.com

Product Code: Ethanol in Nitrogen Part Number: BAC

#### Section 2: Hazards Identification



Hazard Classification: Gases Under Pressure

Hazard Statements: Contains gas under pressure; may explode if heated

Precautionary Statements

Storage: Protect from sunlight. Store in well-ventilated place.

#### Section 3: Composition/Information on Ingredients

	CAS#	Concentration
Ethano!		5-500 ppm
Nitrogen	7727-37-9	Balance

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	Methods for Gleanup	Other Information
Ethanol	Small spills: Absorb with sand or other noncombustible material. Collect spilled material in appropriate container for disposal. Large spills: Dike for later disposal.	
Nitrogen	N/A	N/A

## Section 7: Handling and Storage

	Handling	Storage
Ethanol	Store and use with adequate ventilation. Firmity secure cylindors upright to keep them from falling or being knocked over. Store valve protection cap firmity in place by hand. Store only where temperature will not exceed 125°F (25°C). Store full and empty cylindors separately. Use a first-h, first-out inventory system to prevent storing full cylindors for long particles.	Do not get liquid in eyes, on skin, or clothing. Protect cylinders from damage. Use a suitable hand truck to move cylinders; do not drag, rod, slida, or drop. Open valve slowly. Close cylinder valve after each use; keep closed even when empty. If valve is hard to open, discontinue use and contact your supplier.
Nitrogen	Store and handle in accordance with all current regulations and standards. Subject to storage regulations: U.S. OSHA 29 CFR 1910.101.	Keep separated from incompatible substances.

## Section 8: Exposure Controls/Personal Protection

	Exposure Guidelines
	ETHYL ALCOHOL, 100%: ETHYL ALCOHOL (ETHANOL): 1000 ppm (1900 mg/m3) OSHA TWA 1000 ppm ACGIH TWA 1000 ppm (1900 mg/m3) NIOSH recommended TWA 10 hour(s)
Nitrogen	NITROGEN, COMPRESSED GAS: NITROGEN: ACGIN (sumple amphysiant)

## **Engineering Controls**

	Eye Protection	Skin Protection	Respiratory Protection
Ethanol	Wear splash resistant safety goggles. Provide an emergency eye wash fountain and quick drench shower in the immediate work area.	Wear appropriate chemical resistant clothing.	Any supplied-air respirator with full facepiece and operated in a pressure-demand or other positive- pressure mode in combination with a separate escape supply.
Nitrogen	Eye protection not required, but recommended.	Protective clothing is not required.	Respiratory protection may be needed for frequent or heavy currosure.

General Hygiene considerations

- Avoid breathing vapor or mist
   Avoid contact with eyes and skin
   Wash thoroughly after handling and before eating or drinking

# Section 9: Physical and Chemical Properties

	Boiling Point	Free		Vapor Pressure	Vapor Density	Specifi Gravity		y PH		Odor Threshold	Eva Ret	poration 6	Viscosity		
Nitrogen	Not flamn	nable	Not	available	Not available		Not available North		Norflammabi	orflammable		Nonflammable		Nonflemmable	
Ethanol	55 F (13 (	C)	IB				685 F (363 C)		0.1	0.19		0.033			
	Flash Po	int	Flan	nmability	Partition Coefficie		Autolgnition Temperature			per Explosiv nits	9	Lower E	xplosive		
Nitrogen	Gas		Ciea		Coloriess	N/A		Gas		Odorless	_	Tasteless	1		
Ethanol	Liquid		Cine		Coloriass	N/A		Volable	Inquid	Pleasant o		Bruning (99)	8		

Physical State | Appearance | Color | Change in Appearance | Physical Form | Odor | Tasta

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	Chemical Substance	Chemical Family	Trade Names
Ethanol	ETHYL ALCOHOL, 100%	hydroxyls, eliphatic, alcohols, eliphatic	ETHANOL; ETHYL AL COHOL; 4; AL COHOL; AL COHOL ANHYDROUS; ALGRAIN; ANIYDROL; Absolute alcohol; Anhydrous ethanol; Ethanol denatured; Fermentation alcohol; Grain alcoho; 4-Hydroxyethane; Methyl carbind; Ethyl alcohol enhydrous; Absolute ethanol; Denatured ethanol; ETHYL HYDRATE; ETHYL HYDROXIDE; JAYSOL; TECSOL; STCC 4090159; IM 170: C23460
Nitrogen	NITROGEN, COMPRESSED GAS	Inorganic gases	DIATOMIC NITROGEN; DINITROGEN; NITROGEN; NITROGEN-14; NITROGEN GAS; UN 1066; N2

# Section 4: First Aid Measures

	Skin Contact	Eye Contact	Ingestion	inhalation	Note to Physicians
Ethanol	Wesh skin with soep and water for at least 15 minutes while removing contaminated clothing and shoes. Get medical attention, if needed. Thoroughly clean and dry contaminated clothing and shoes before reuse.	Flush eyes with plenty of water for at least 15 minutes. Then get immediate medical attention.	Context local poison control center or physician immediately. Never make an unconscious parson vomit or drink fluids. When vamining occurs, keep head lower than hips to help prevent aspiration. If person is unconscious, turn head to side. Got medical attention immediately.	If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing. Get immediate medical attention.	For ingestion consider gastric lavage.
Nitrogen	Wash exposed skin with scap and water.	Flush eyes with plenty of water.	If a large amount is swallowed, get medical attention.	If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing. If breathing is difficult, oxygen should be administered by qualified personnel. Get immediate medical attention.	For inhalation, consider oxygen.

#### Section 5: Fire Fighting Measures

	Suitable Extinguishing Media	Products of Combustion	Protection of Firefighters
Ethanol	Alcohol resistant losm, carbon dioxido, regular dry charnical, weder, alcohol resistant foam large fires; Uso alcohol-resistant foam or flood with fine water spray.	Carbon monoxide, carbon dioxide, and toxic and imitating furnes	<ul> <li>Any supplied-air respirator with full flampings and operated in a pressure-demand or other positive-pressure mode in combination with a separato oscape supply.</li> <li>Any supplied-air respirator with full flampings and operated in a pressure-demand or other positive-pressure mode in combination with a separate escape supplied.</li> </ul>
Nitrogen	Non-flammable. Use suitable extinguishing media for surrounding fire. Cylinders may rupture or explode if supposed to heat.	Non-flammable	<ul> <li>Respiratory protection may be needed for frequent or heavy exposure.</li> </ul>

#### Section 6: Accidental Release Measures

	Personal Precautions	Environmental Precautions	Methods for Containment
Ethanol	Keep unnecessary people away, isolate hazard area and deny entry. Stay upwind and keep out of low areas.	Avoid heat, flames, sparks and other sources of ignition.	Stop leak if possible without personal risk, Reduce vapors with water spray, Remove sources of ignition.
Nitrogen	Keep unnecessary people away, isolate hazard area and deny entry. Stay upwind and keep out of low areas.	No significant effects from contamination expected.	Stop leak if possible without personal risk.

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	Boiling Point	Freezing Point	Vapor Pressure	Vapor Density	Specific Gravity	Water Solublility	pН	Odor Threshold	Evaporation Rate	Viscosity
Ethanol	172 F (78 C)	-179 F (- 117 C)	40 mmHg @ 19 C	1.59 (Air=1)	0.7693	Soluble	Not avaitable	5-10 ppm	1.4 (carbon telrachlorids=1)	1.17 mPa.s (1.17 centpoises) @ 20 C; 1.074 mPa.s (1.074 centipoises @ 25 C
Nitrogen	-321 F (-195 C)	-346 F (- 210 C)	760 mmHg @ -196 C	0.967 (Air=1)	Not applicable	1.6% @ 20 C	Not applicable	Not available	Not applicable	0.01787 cP @ 27 C

	Molecular Weight	Molecular Formula	Density	Weight per Gallon	Volatility by Volume	Volatility	Solvent Solubility
Ethanol	46.07	C-H3-C-H2-O- H	Not available	Not available	Not available	1	Soluble: Senzene, ether, acetone, chloroform, methanol, anj anic solvents
Nitrogen	28.0134	N2	1.2506 g/L	Not available	100%	1	Soluble: Liquid ammonia

## Section 10: Stability and Reactivity

	Stability	Conditions to Avoid	Incompatible Materials		
Ethanol	Stable at normal temperatures and pressure.	Stable at normal temperatures and pressure.	Halo carbons, metals, metal salts, oxidizing materials, halogens, peroxides acids metal oxides bases combustible materials		
Nitrogen	Stable at normal temperatures and pressure.	Stable at normal temperatures and pressure.	Metals, oxidizing materials		

	Hazardous Decomposition Products	Possibility of Hazardous Reactions		
Ethanol	Oxides of carbon	Will not polymarize.		
Kitrogen	Oxidee of nitrogen	Will not polymerize.		

#### Section 11: Toxicology Information

Oral LD50 Dermat LD50		Dermat LD50	inhalation		
Ethanol	7 gm/kg orel- rat LD50	LD50 (dermal, rebbit): greater than 15800 mg/kg (cited as greater than 20 mL/kg); at 20 mL/kg, 1/4 rebbits died	Irritation, difficulty breathing, headache, drowsiness, symptoms of drunkenness		
Nitrogen	Not available	Not evailable	Nausea, vomiting, difficulty breathing, headache, drowsiness, dizziness, tingling sensation, loss of coordination, convulsions come		

	Eye Irritation	Skin Irritation	Sensitization
Ethanol	Irritation, tearing	Mild irritation, rash	Respiratory tract imitation, skin imitation, eye imitation, liver damage, central nervous system decression
Nitrogen	Contect with rapidly expanding gas	No information on significant adverse effects	Difficulty breathing

	Carcinogenicity	Mutagenicity	Reproductive Effects	Developmental Effects
Ethanol	NTP: Known Human Carcinogen (Alcoholic beverages); IARC: Human Sufficient Evidence, Group 1 (Alcoholic beverages), Animal Inadequate Evidence; ACGIH: A4-Not Classifiable as a Human Carcinogen	Available.	Available.	No data
Nitrogen	Not hazardous	Not available	Not available	No data